## **DIPLOMA - COMMON ENTRANCE TEST-2016**

CD	COURSE	DAY: SUNDAY		
CR	CERAMICS	TIME: 10.00 a.m. to 1.00 p.m.		

MAXIMUM MARKS	TOTAL DURATION	MAXIMUM TIME FOR ANSWERING
180	200 MINUTES	180 MINUTES

MENTION YOUR	<b>QUESTION BOOKLET DETAILS</b>				
DIPLOMA CET NUMBER	VERSION CODE	SERIAL NUMBER			
	A - 1	100301			

#### DOs:

- Check whether the Diploma CET No. has been entered and shaded in the respective circles on the OMR answer sheet.
- 2. This Question Booklet is issued to you by the invigilator after the 2<sup>nd</sup> Bell i.e., after 09.50 a.m.
- 3. The Serial Number of this question booklet should be entered on the OMR answer sheet and the respective circles should also be shaded completely.
- 4. The Version Code of this question booklet should be entered on the OMR answer sheet and the respective circles should also be shaded completely.
- 5. Compulsorily sign at the bottom portion of the OMR answer sheet in the space provided.

#### DON'Ts:

- 1. THE TIMING AND MARKS PRINTED ON THE OMR ANSWER SHEET SHOULD NOT BE DAMAGED/MUTILATED/SPOILED.
- 2. The 3rd Bell rings at 10.00 a.m., till then;
  - Do not remove the paper seal / polythene bag of this question booklet.
  - Do not look inside this question booklet.
    - Do not start answering on the OMR answer sheet.

### IMPORTANT INSTRUCTIONS TO CANDIDATES

- This question booklet contains 180 (items) questions and each question will have one statement and four answers. (Four different options / responses.)
- 2. After the 3<sup>rd</sup> Bell is rung at 10.00 a.m., remove the paper seal / polythene bag of this question booklet and check that this booklet does not have any unprinted or torn or missing pages or items etc., if so, get it replaced by a complete test booklet. Read each item and start answering on the OMR answer sheet.
- 3. During the subsequent 180 minutes:
  - Read each question (item) carefully.
  - Choose one correct answer from out of the four available responses (options / choices) given under each question / item. In case you feel that there is more than one correct response, mark the response which you consider the best. In any case, choose only one response for each item.
  - Completely darken / shade the relevant circle with a BLUE OR BLACK INK BALL POINT PEN
    against the question number on the OMR answer sheet.

Correct	Method	of	shading	the	circle	ōn	the	OMR	answer	sheet	ís	as	shown	below	:
					<b>A</b>		<b>D</b> (1	<b>D</b>							

- Use the space provided on each page of the question booklet for Rough Work. Do not use the OMR answer sheet for the same.
- 5. After the last Bell is rung at 1.00 p.m., stop marking on the OMR answer sheet and affix your left hand thumb impression on the OMR answer sheet as per the instructions.
- 6. Hand over the OMR ANSWER SHEET to the room invigilator as it is.
- 7. After separating the top sheet (KEA copy), the invigilator will return the bottom sheet replica (Candidate's copy) to you to carry home for self-evaluation.
- 8. Preserve the replica of the OMR answer sheet for a minimum period of ONE year.



CR-A1

# APPLIED SCIENCE

1	An ex	xample of basic S.I. unit is		
	(A)	Newton	(B)	Joule
	(C)	Ampere	(D)	Watt
2.	The	prefix used for $10^{+2}$ is		
	(A)	hecta	(B)	centi
	(C)	pico	(D)	peta
3.	An e	xample of dimensionless physical	quanti	ty is
	(A)	surface tension	(B)	strain
	(C)	impulse	(D)	period
		1 to C C 1 Cillian hadron	eoduoll	ae it falls
4.		velocity of a freely falling body gr		
	(A)	decreases	(B)	increases
	(C)	remains same	(D)	increases and then decreases
5.		ain scale is divided into half mm at count of cm.	and hav	ving a vernier containing 20 divisions has a
	(A)	$2.5 \times 10^{-2}$	(B)	$0.5 \times 10^{-2}$
	(C)	$0.025 \times 10^{-2}$	(D)	$0.25 \times 10^{-2}$
6.	For	a particular mass of the moving bo	ody, its	s friction is minimum when it is
	(A)	sliding	(B)	static
	(C)	rolling	(D)	dragged
		Snaca	For Ro	ough Work

		Space F	or Rou	igh Work	
	(C)	Law of triangle of forces.	(D)	Law of polygon of force	S.
	(A)	Law of parallelogram of forces.	(B)	Lami's theorem.	
12.	Tow	ring of a boat by two forces is an ill	ustrati	on of	
	(C)	equal to other	(D)	obtuse	
	(A)	biggest	(B)	smallest	
11.		en three forces acting at a point are anys angle.	in equ		e to biggest force is
	(C)	$kg m s^{-2}$	(D)	$kg m s^{-1}$	
	(A)	kg m	(B)	$kg m^{-1}s^{-1}$	
10.	The	S.I. unit of momentum is			
	(C)	Lami's theorem	(D)	Polygon of forces	
	(A)	Parallelogram of forces	(B)	Triangle of forces	
9.	То с	check the equilibrium of five coplan	ar con	current forces, we use law	of
	(C)	9 m/s	(D)	7.5 m/s	
	(A)	4.9 m/s	(B)	18 m/s	
8.		Force of $1.5 \times 10^{-2}$ N acts for 3 seconds. The final velocity of the body is	ids on	a body of mass 0.05 kg mo	oving with velocity
	(C)	variable velocity	(D)	variable acceleration	
	(A)	constant velocity	(B)	constant acceleration	
7.	All	equations of motion hold good unde	er the	condition of	

13.		esultant force on the bod		eously making an angle (	
	(A)	8 N	(B)	4 N	
	(C)	7 N	(D)	49 N	
14.	Dime	ensional formula for stres	s is		
	(A)	$[LM^{-1}T^{-2}]$	(B)	$[L^{-1}MT^{-2}]$	
	(C)	$[L^{-1}M^{-1}T]$	(D)	$[L^2M^{-1}T^{-2}]$	
15.	The	pull in the bicycle chain	is an example of		
	(A)	tensile stress	(B)	volume stress	
	(C)	shear stress	(D)	shear strain	
16.	Viso	osity of water at 20 °C in	centipoise is		
	(A)	1.792	(B)	0.650	
	(C)	1.005	(D)	0.470	
17.	Dim	ensional formula of surf	ace tension is		
	(A)	[LMT <sup>-2</sup> ]	(B)	$[L^2MT^{-2}]$	
	(C)	$[LM^{-1}T^{-2}]$	(D)	$[L^0MT^{-2}]$	
18.	A st	eel needle can be floated	on the surface of	f water because of the	
	(A)	density of steel is great	ter than water		
	(B)	density of steel is less	than water		
	(C)	surface tension			

(D) viscosity

19.		st on the bottom of the container hant of 6 m is	ving a	a base area of 10 m <sup>2</sup> filled with water to a
	(A)	$60 \times 10^2 \text{ N}$	(B)	$58.8 \times 10^4 \text{ N}$
	(C)	60.8 N	(D)	600 N
20.	Keep	oing the temperature constant, if the	press	ure of the gas is doubled its volume
	(A)	remains constant	(B)	doubles
	(C)	reduces to one fourth	(D)	reduces to half
21.	Heat	transfer in the absence of the media	um is	
21.	(A)	conduction	(B)	convection
	(C)	radiation	(D)	absorption
22.	Zero	of absolute scale of temperature is	at	
	(A)	0 °C	(B)	100 °C
	(C)	273 °C	(D)	−273 °C
23.	Ripp	les on water surface is an example	of	
	(A)	electromagnetic waves	(B)	transverse waves
	(C)	waves travelling in space	(D)	longitudinal waves
24.	The	time interval between two consecut	ive wa	axing and waning of sound waves is
	(A)	beat period	(B)	wave period
	(C)	beat frequency	(D)	wave frequency
		Space Fo	r Rou	gh Work

- 25. S.I. unit of intensity of sound is
  - (A) watt per square meter
- (B) watt per meter

(C) watt square meter

- (D) watt meter
- 26. The study of characteristics of buildings with reference to sound is
  - (A) resonance

(B) interference

(C) echo

- (D) acoustics
- 27. The distance travelled by the disturbance in the medium for one complete oscillation is
  - (A) wave velocity

(B) wavelength

(C) wave frequency

- (D) wave amplitude
- 28. Momentum of a photon is given by
  - (A)  $P = \frac{\lambda}{h}$

(B)  $P = \frac{h}{\lambda}$ 

(C)  $P = \lambda h$ 

- (D)  $P = \lambda^2 h$
- 29. The velocity of sound in case of liquids is given by
  - (A)  $\sqrt{\frac{d}{k}}$

(B) √kd

(C)  $\sqrt{\frac{k}{d}}$ 

A-1

- (D)  $\sqrt{\frac{d^2}{k}}$
- 30. A tuning fork vibrating in air is an example of
  - (A) damped free vibrations
- (B) resonant vibrations
- (C) undamped free vibrations
- (D) forced vibrations

- 31. Raman lines are
  - (A) unpolarised

(B) polarised

(C) diffracted

- (D) reflected
- 32. A crystal which has two optic axes is
  - (A) calcite

(B) quartz

(C) mica

- (D) glass
- 33. Electron microscope is used to
  - (A) study virus and bacteria
  - (B) view three dimensional images
  - (C) automatic switching on and off of street-lights
  - (D) electronic industry for soldering
- 34. Which of the following statements is correct in case of  $\gamma$ -rays?
  - (A) Penetrating power is less than  $\beta$ -rays.
  - (B) Penetrating power is less than  $\alpha$ -rays.
  - (C) Penetrating power is very high.
  - (D)  $\gamma$  particles are nothing but electrons.
- 35. For destructive interference of light the path difference should always be
  - (A)  $(2n+1)\frac{\lambda}{2}$

(B)  $\frac{n\lambda}{2}$ 

(C)  $(2n+1) \frac{\lambda}{3}$ 

(D)  $n\lambda$ 

- 36. The resultant intensity of interference of two monochromatic waves having same amplitude and constant phase difference equal to  $\phi$  is
  - (A)  $2a \cos\left(\frac{\phi}{2}\right)$

(B)  $4a^2\cos^2\left(\frac{\phi}{2}\right)$ 

(C)  $4a^2\cos\left(\frac{\phi}{2}\right)$ 

- (D)  $4a \cos^2\left(\frac{\phi}{2}\right)$
- 37. For two objects to be just resolved, the principle maximum should be on
  - (A) first maximum

(B) second maximum

(C) first minimum

- (D) second minimum
- 38. Resolving power of microscope is given by
  - (A)  $\frac{\lambda}{2n\sin\theta}$

(B)  $\frac{n}{2\lambda \sin \theta}$ 

(C)  $\frac{2\lambda\sin\theta}{n}$ 

- (D)  $\frac{2n \sin \theta}{\lambda}$
- 39. In case of acids, the concentration of H<sup>+</sup> ions is
  - (A) more than  $10^{-7}$  g ions/litre.
  - (B) less than  $10^{-7}$  g ions/litre.
  - (C) equal to  $10^{-7}$  g ions/litre.
  - (D) between  $10^{-7}$  g ions/litre and  $10^{-14}$  g ions/litre.
- 40. Corrosion of metal can be prevented by keeping it in
  - (A) acidic medium

(B) basic medium

(C) neutral medium

(D) moisture

# PART – B APPLIED MATHEMATICS

41. The value of the determinant 
$$A = \begin{bmatrix} 1 & 1 & 1 \\ 3 & 3 & 3 \\ 4 & 5 & 6 \end{bmatrix}$$
 is

(A) 1

(B) 3

(C) -2

(D) 0

42. The value 'x' by Cramer's rule in 
$$3x + 2y = 4$$
 and  $x - 2y = 8$  is

(A) 12

(B) 3

(C) - 13

(D) 15

43. If 
$$A = \begin{bmatrix} 2 & -3 \\ 1 & 5 \end{bmatrix}$$
  $B = \begin{bmatrix} 1 & 2 \\ 4 & -3 \end{bmatrix}$ , then  $A + 2B$  is

 $(A) \begin{bmatrix} 4 & 1 \\ 9 & -1 \end{bmatrix}$ 

(B)  $\begin{bmatrix} 4 & 1 \\ 9 & 1 \end{bmatrix}$ 

(C)  $\begin{bmatrix} 3 & -1 \\ 5 & 2 \end{bmatrix}$ 

(D)  $\begin{bmatrix} 3 & 1 \\ 5 & 2 \end{bmatrix}$ 

44. If 
$$A = \begin{bmatrix} 2 & 3 & 4 \\ -2 & x & -4 \\ -5 & 6 & 7 \end{bmatrix}$$
 is singular, then the value of x is

(A) -3

(B) 3

(C)  $\frac{1}{3}$ 

(D)  $\frac{-1}{3}$ 

- **45.** The characteristic roots of the matrix  $A = \begin{bmatrix} 1 & 4 \\ 3 & 2 \end{bmatrix}$  is
  - (A) 5, 2

(B) -5, -2

(C) 5, -2

- (D) -5, 2
- 46. If  ${}^{n}C_{16} = {}^{n}C_{3}$ , then the value of n is
  - (A) -19

(B) 19

(C) 13

- (D) -13
- 47. The last term in the expansion of  $\left(3x^2 + \frac{1}{2x^2}\right)^4$  is
  - $(A) \quad \frac{1}{8x^8}$

(B)  $\frac{1}{16x^8}$ 

(C)  $81 x^8$ 

- (D)  $12 x^8$
- 48. The unit vector of  $\vec{a} = 2i 3j + 4k$  is
  - $(A) \quad \frac{2i-3j+4k}{\sqrt{29}}$

 $(B) \quad \frac{2i-3j+4k}{\sqrt{11}}$ 

(C)  $\frac{2i-3j+4k}{\sqrt{3}}$ 

- (D)  $\frac{\sqrt{29}}{2i-3j+4k}$
- 49. If  $\vec{a} = i 4j + 3k$  and  $\vec{b} = -2i + j + 6k$ , then the projection of  $\vec{a}$  on  $\vec{b}$  is
  - (A)  $\frac{24}{\sqrt{41}}$

(B)  $\frac{12}{\sqrt{26}}$ 

(C)  $\frac{-12}{\sqrt{41}}$ 

(D)  $\frac{12}{\sqrt{41}}$ 

**50.** The area of triangle whose two sides are  $\vec{a} = 3i + 4j + k$  and  $\vec{b} = 5i + 6j + 2k$  is

(A) 3 sq. units

(B)  $\frac{1}{2}$  sq. units

(C)  $\frac{3}{2}$  sq. units

(D)  $\frac{9}{2}$  sq. units

51. The simplification of  $\frac{1}{1+\sin\theta} + \frac{1}{1-\sin\theta}$  is

(A)  $2\cos^2\theta$ 

(B)  $2 \sec^2 \theta$ 

(C)  $\tan^2 \theta$ 

(D)  $2 \csc^2 \theta$ 

52. The value of  $\tan^2 30^\circ + \sin^2 45^\circ + \cos^2 90^\circ + \cos^2 60^\circ$  is

(A)  $\frac{4}{3}$ 

(B)  $\frac{13}{12}$ 

(C)  $\frac{13}{24}$ 

(D)  $\frac{25}{12}$ 

53. The simplification of  $\frac{\sin (180^{\circ} - A) \cos (360^{\circ} - A)}{\tan (90^{\circ} + A) \sin (-A)}$  is

(A) sin A

(B) cosec A

(C) - sin A

(D) - cosec A

54. If  $\cos A = \frac{-3}{5}$  where  $90^{\circ} < A < 180^{\circ}$ , then the value of  $\cot A$  is

(A)  $\frac{3}{4}$ 

(B)  $\frac{4}{3}$ 

(C)  $\frac{-3}{4}$ 

(D)  $\frac{-4}{3}$ 

55. The value of cos 105° is

 $(A) \quad \frac{\sqrt{3}-1}{2\sqrt{2}}$ 

 $(B) \quad \frac{\sqrt{3}+1}{2\sqrt{2}}$ 

 $(C) \quad \frac{2\sqrt{2}}{1-\sqrt{3}}$ 

(D)  $\frac{1-\sqrt{3}}{2\sqrt{2}}$ 

56. If  $\tan \frac{A}{2} = \frac{1 - \cos A}{\sin A}$ , then the value of  $\tan 22 \frac{1^{\circ}}{2}$  is

(A)  $\sqrt{2} + 1$ 

(B)  $1 - \sqrt{2}$ 

(C)  $\sqrt{2} - 1$ 

(D)  $-1 - \sqrt{2}$ 

57. The value of  $\cos 5x \cdot \cos 3x$  is

(A)  $\cos 8x + \cos 2x$ 

- (B)  $\frac{1}{2} (\cos 8x + \cos 2x)$
- (C)  $\frac{1}{2} \left( \sin 8x + \sin 2x \right)$
- (D)  $\frac{1}{2} (\cos 8x \cos 2x)$

58. The simplified value of  $\tan^{-1}\left(\frac{1}{2}\right) + \tan^{-1}\left(\frac{1}{3}\right)$  is

(A)  $\frac{\pi}{4}$ 

(B)  $\frac{\pi}{3}$ 

(C) 1

(D)  $\tan^{-1}\left(\frac{1}{7}\right)$ 

59. Distance of a point P(-2, 5) from the origin is

(A)  $\sqrt{29}$ 

(B)  $\sqrt{21}$ 

(C)  $\sqrt{3}$ 

(D) 29

60. The co-ordinates of the point which divides the line joining the points A (8, 3) and B(-5, 6) in the ratio of 2:3 externally is

(A) (-34, -3)

(B) (34, 3)

 $(C) \quad \left(\frac{14}{5}, \frac{21}{5}\right)$ 

(D) (34, -3)

61. The area of triangle with the vertices (5, 3), (4, 6) and (5, 8) is

(A)  $\frac{15}{2}$  sq. units

(B) 15 sq. units

(C)  $\frac{5}{2}$  sq. units

(D)  $\frac{45}{2}$  sq. units

62. The slope of the line making an angle  $150^{\circ}$  with the x-axis is

 $(A) \quad \frac{-1}{\sqrt{3}}$ 

(B)  $\frac{1}{\sqrt{3}}$ 

(C)  $\sqrt{3}$ 

(D)  $-\sqrt{3}$ 

63. The two point form of a straight line is

(A)  $y - y_1 = m(x - x_1)$ 

(B)  $\frac{y-y_1}{x-x_1} = \frac{y_2-y_1}{x_2-x_1}$ 

(C)  $\frac{y}{x} = \frac{y_2 - y_1}{x_2 - x_1}$ 

(D)  $\frac{y-y_2}{x-x_2} = \frac{y_2-y_1}{x_2-x_1}$ 

64. The equation of straight line perpendicular to 2x + 5y - 8 = 0 and passing through (-1, 2) is

(A) 2x + 5y + 9 = 0

(B) 5x - 2y + 1 = 0

(C) 5x - 2y + 9 = 0

(D) 5x + 2y - 9 = 0

**65.** The value of  $\lim_{x \to 3} \frac{2x^2 - 7x + 3}{2x - 6}$  is

(A) 3

(B)  $\frac{2}{5}$ 

(C)  $\frac{5}{2}$ 

(D) 5

- **66.** The value of  $\lim_{x\to 0} \frac{\sqrt{1-\cos x}}{x}$  is
  - (A)  $\frac{1}{\sqrt{2}}$

(B)  $\sqrt{2}$ 

(C)  $\frac{1}{2}$ 

- (D) 1
- 67. If  $y = e^x (\cos x \sin x)$ , then  $\frac{dy}{dx}$  is
  - (A)  $2e^x \cos x$

(B)  $-2e^x \cos x$ 

(C)  $2e^x \sin x$ 

- (D)  $-2e^x \sin x$
- 68. If  $x + y = \log x + \log y$ , then  $\frac{dy}{dx}$  at x = -1 and y = 2 is
  - (A)  $-\frac{1}{4}$

(B) -4

(C) 4

- (D)  $\frac{1}{2}$
- 69. If  $x = a \cos^2 \theta$  and  $y = b \sin^2 \theta$ , then  $\frac{dy}{dx}$  is
  - (A)  $\frac{-b}{a}$

(B)  $\frac{b}{a}$ 

(C)  $\frac{a}{b}$ 

- (D)  $\frac{-a}{b}$
- 70. The second derivative of  $y = \log \left(\frac{1}{x}\right)$  is
  - (A) x

(B) 1

(C)  $\frac{1}{x^2}$ 

(D)  $\frac{-1}{x^2}$ 

71. The equation of normal to the curve  $y = (2x + 1)^2$  at (-2, 0) is

(A) 
$$x - 16y + 2 = 0$$

(B) 
$$x - 12y + 2 = 0$$

(C) 
$$x + 16y + 2 = 0$$

(D) 
$$x + 12y + 2 = 0$$

72. The maximum value of the function  $y = 2x^3 + 3x^2 - 36x$  is

$$(A) - 44$$

(B) 
$$-30$$

$$(D) - 81$$

73. The value of  $\int \sin 3x \cos 2x \, dx$  is

(A) 
$$\frac{-1}{2} \left[ \frac{\cos 5x}{5} + \cos x \right] + C$$

(B) 
$$\frac{1}{2} \left[ \frac{-\cos 5x}{5} + \cos x \right] + C$$

(C) 
$$\frac{1}{2} \left[ \frac{\cos 5x}{5} + \cos x \right] + C$$

(D) 
$$\frac{-1}{2} [\cos 5x + \cos x] + C$$

74. The value of  $\int x^2 \sin(2x^3) dx$  is

(A) 
$$\frac{-\cos(2x^3)}{6} + C$$

(B) 
$$\frac{-\cos(2x^3)}{3} + C$$

(C) 
$$12x^3\cos(2x^3) + C$$

(D) 
$$\frac{\cos(2x^3)}{6} + C$$

75.  $\int \log x \, dx$  is

(A) 
$$\frac{1}{x}$$
 + C

(B) 
$$\frac{1}{x} - x + C$$

(C) 
$$x \log x + x + C$$

(D) 
$$x \log x - x + C$$

- 76. The value of  $\int_{0}^{\pi^2} \sqrt{1+\sin 2x} \, dx$  is
  - (A) 0

(B) 1

(C) 2

(D) -2

- 77.  $\int_{0}^{1} \frac{x}{1+x^4}$  is
  - (A)  $\frac{\pi}{4}$

(B)  $\frac{\pi}{8}$ 

(C)  $\frac{-\pi}{8}$ 

- (D)  $\frac{-\pi}{4}$
- 78. The area formed by the curve  $y = (2x + 1)^3$  between the ordinates x = -1 and x = 1 is
  - (A)  $\frac{41}{4}$  sq. units

(B) 2 sq. units

(C) 20 sq. units

- (D) 10 sq. units
- 79. The order and degree of differential equation  $\left[1+\left(\frac{dy}{dx}\right)^4\right]^{2/3} = \frac{d^2y}{dx^2}$  is
  - (A) order 2 and degree 3
- (B) order 2 and degree 1
- (C) order 1 and degree 2
- (D) order 1 and degree 4
- 80. The solution of differential equation  $\sec^2 x \tan y \, dx + \sec^2 y \tan x \, dy = 0$  is
  - (A)  $\tan^2 x + \tan^2 y = C$

(B)  $\tan x + \tan y = C$ 

(C)  $\tan x \tan y = C$ 

(D)  $x + y + \log(\sec x \sec y) = C$ 

## **PART-C**

# **CERAMICS**

81.	Mark	ole stone is			
	(A)	Igneous rock		(B)	Sedimentary rock
	(C)	Metamorphic rock		(D)	Not a rock
82.	Hack	cly is a term used for			
	(A)	Form		(B)	Fracture
	(C)	Hardness		(D)	Cleavage
83.	The	depth of interior of the ear	th is estin	nated	about
	(A)	2900 km		(B)	1000 km
	(C)	5000 km		(D)	6370 km
84.	Gyp	sum crystalizes under	syste	m.	
	(A)	Monoclinic		(B)	Hexagonal
	(C)	Tetragonal		(D)	Isometric
85.	Rocl	ks are defined as	_ mineral:	s.	
	(A)	Pure soil		(B)	Impure soil
	(C)	Aggregate		(D)	None of these
86.	Hard	lness of Quartz on Mohs s	cale is		
	(A)	7		(B)	8
	(C)	9		(D)	10
			Space F	or Roi	igh Work

7.	Insid	le the crust there are rock layers kno	wn as	
	(A)	Hydrosphere	(B)	Mesosphere
	(C)	Atmosphere	(D)	Lithosphere
88.	Two	adjacent faces meet to form a/an		
	(A)	External angle	(B)	Internal angle
	(C)	Straight angle	(D)	No angle
<b>39.</b>	The	poorest property of the Ceramics is		
,,,	(A)	Hard	(B)	Dense
	(C)	Brittle	(D)	Tough
90.		is used as frothing agent.		
vu.	(A)	Pine oil	(B)	Grease
	(A) (C)	Wax	(D)	Varnish
91.	POF	is prepared from mineral.		
	(A)	Calcite	(B)	Baddylite
	(C)	Gypsum	(D)	Cryolite
92.	Exa	mple for alumina mineral is		
	(A)	Diaspore	(B)	Gibbsite
	(C)	Bauxite	(D)	All of these
93.	Por	celain is a body.		
	(A)	Vitreous	(B)	Semi-vitreous
	(C)	Non-vitreous	(D)	Opaque
				b wy I

94.		decoration method is commo	nly us	ed for tiles.		
	(A)	Etching	(B)	Sand blasting		
	(C)	Screen printing	(D)	Gunning		
95.	The	hardness of tale on Mohs scale is				
	(A)	1	(B)	2		
	(C)	3	(D)	4		
96.	Thic	k sanitary wares are dried in				
	(A)	Rotary drier	(B)	Humidity drier		
	(C)	Spray drier	(D)	Mangle drier		
97.	Di-e	lectric strength is the property of				
	(A)	Glazed tile	(B)	Sewer pipes		
	(C)	Bone china ware	(D)	Porcelain insulator		
98.	Whi	ch of the following affects the colo	ur of t	he ware ?		
	(A)	Colouring agent	(B)	Kiln atmosphere		
	(C)	Firing temperature	(D)	All of these		
99.	Com	nmon impurities in calcined Alumin	na is			
	(A)	Magnesia, Potash, Soda				
	(B)	Titanium, Silica, Iron oxide				
	(C)	Calcium oxide, Barium oxide, Ni	ckel o	xide		
	(D)	Cobalt oxide, Antiomony oxide,	Chrom	ium oxide		
					Y 17	

00.	Basic	refractories which consists of bas	ic mate	erial have resistance to
	(A)	Basic slag	(B)	Acidic slag
	(C)	Both basic and acidic slag	(D)	High temperature
1.	Chen	nical formula of diaspore and bach	mite is	
	(A)	$Al_2O_3$		$Al_2O_3 \cdot H_2O$
	(C)	$Al_{25}O_3$	(D)	$3Al_2O_3 \cdot H_2O$
2.	Afte	r sharping the drying temperature o	of chro	me/mag refractory bricks/blocks is
		95 – 100 °C		100 − 105 °C
	(C)	150 – 250°C	(D)	450 – 550 °C
3.	The	chemical composition of chrome o	re is	
	(A)	$Cr_2O_3$	(B)	CrO <sub>2</sub>
	(C)	CrO	(D)	$CrO_3$
14.	Exar	nple of heat insulation refractory n	nateria	l is
	(A)	Mica	(B)	Porcelain
	(C)	Stoneware	(D)	Glass
)5.		manufacture of refractory bricks nning of	& blo	cks in England and USA was started in the
	(A)	19 <sup>th</sup> Century	(B)	18 <sup>th</sup> Century
	(C)	17 <sup>th</sup> Century	(D)	15 <sup>th</sup> Century
)6.	The	main raw material in the manufact	ure of	crude fused Alumina is
	(A)	Raw Bauxite	(B)	Calcined Bauxite
	(C)	Kyanite	(D)	Silliminate

		Sman F	on Do	ugh Work			
	(C)	Vapour deposition	(D)	Polymerization			
		osit is known as Precipitation	(B)	Gelation			
113.	The condensation of an element or a compound from the vapour state, forming a solid						
	(C)	Iso-static pressing	(D)	Tape casting			
	(A)	Extrusion	(B)	Slip casting			
112.	Plas	tic clay bodies have traditionally be	en for	med by			
	(C)	Biology	(D)	Zoology			
	(A)	Petrology	(B)	Rheology			
111.		is the science of deformation	and fl	ow.			
	(C)	69.8%	(D)	74.0%			
	(A)	52.4%	(B)	60.5%			
110.	The	packing density of tetrahedral confi	gurati	ons is			
	(C)	Plasticizer	(D)	Preservative			
	(A)	Foaming agent	(B)	Antifoaming agent			
109.	Sodi	um alkyl sulfate is an example for					
	(C)	3.1 – 3.5	(D)	3.5 – 4			
	(A)	1.5 - 1.9	. ,	2.5 - 2.8			
108.	Carb	on refractories specific gravity is					
	(C)	Fire clay	(D)	Graphite			
	(A)	Alumina	(B)	Kyanite			
107.		actory Saggers are manufactured by					

114.	Trim	ming is commonly performed when	the b	ody is in state.	
	(A)	Dry	(B)	Plastic	
	(C)	Slurry	(D)	Leather hard	
115.	In gl	ass SiO <sub>2</sub> is known as			
	(A)	Network former	(B)	Network modifier	
	(C)	Intermediator	(D)	Network slower	
116.	Whi	ch is known as salt cake?			
	(A)	Na <sub>2</sub> SO <sub>4</sub>	(B)	NaCl	
	(C)	NaOH	(D)	$Na_2(PO_4)_2$	
117	Gyn	sum used to prevent			
117.	(A)	Caking	(B)	Stocking	
	(C)	Baking	(D)	None of these	
118.	Mor	e than 95% of all manufactured gla	ss is c	hiefly composed of	
	(A)	Soda-lime glass	(B)	Potash-lime glass	
	(C)	Potash-lead glass	(D)	Optical glass	
119.	The	process of freeing the melt from bu	ıbbles	is known as	
		Fining		Spinning	
	(C)	Milling	(D)	Nilling	
120.	Stra	in in glass can be detected by using			
	(A)	Solenoid strain detector	(B)	Polaroid strain detector	
	(C)	Maxwell strain detector	(D)	Miller strain detector	
					 _

121.	Etchi	ing of glass is done by using		
	(A)	HF acid commercial 60%	(B)	HF acid commercial 30%
	(C)	HF acid commercial 20%	(D)	HF acid commercial 10%
122.	Glas	s breaks only in		
	(A)	Compression	(B)	Tension
	(C)	Suspension	(D)	None of these
123.	The	apparatus used to determine the ther	mal e	xpansion of glass is
	(A)	Screw gauge apparatus	(B)	Dial gage apparatus
	(C)	Hook gauge apparatus	(D)	Littletons apparatus
124.	Glas	s is a of heat.		
	(A)	Poor conductor	(B)	Good conductor
	(C)	Not conductor	(D)	Speed conductor
125.	Cem	ent was invented by Joseph Aspidin	in th	e year
	(A)	1820	(B)	1824
	(C)	1830	(D)	1845
126.	If th	ne cement temperature exceeds 12 property due to partial hydrati		during grinding then such cement exhibit gypsum.
	(A)	False setting	(B)	Good setting
	(C)	Bad setting	(D)	None of these
127.	Cont	tinuous lime-stone calcination kiln i	s cons	structed by using
	(A)	Special refractory bricks	(B)	Refractory bricks
	(C)	Wire cut bricks	(D)	Cement blocks

128.	When	n lime-stone is heated, it will break	down	with heat. We call this reaction as
	(A)	Chemical reaction	(B)	Thermal decomposition
	(C)	Chemical decomposition	(D)	Break down reaction
129.	Air e	ntering the rotary kiln through the c	cooler	s is called
	(A)	Hot air	(B)	Cold compressed air
	(C)	Secondary air	(D)	Supplementary air
130.	The	inclination of rotary kiln is		
	(A)	1 metre height for every 30 metres	lengt	h
	(B)	2 metre height for every 30 metres	lengt	th
	(C)	3 metre height for every 30 metres	lengt	th
	(D)	4 metre height for every 30 metres		
131.	The	chemical formula of gypsum is		
	(A)	CaSO <sub>4</sub> · 2H <sub>2</sub> O	(B)	CaSO <sub>4</sub> ·H <sub>2</sub> O
	(C)	$2\text{CaSO}_4 \cdot \text{H}_2\text{O}$	(D)	$2CaSO_4 \cdot 2H_2O$
132.	In In	ndia portland cement was first manu	factu	red in the year
	(A)	1848	(B)	1912
	(C)	1914	(D)	1904
133.	Whe	en calcite is calcined at 1000 °C it c	hange	es to phase known as
	(A)	Oolite		
	(C)	Belite	(D)	Crystobalite
134.	The	later age's strength in ordinary port	tland (	cement is achived by
		$C_4AF$		$C_3A$
		$C_2S$		C <sub>3</sub> S

A >	_		
A)	Formation of stone	(B)	Formation of clay
C)	Formation of petrolium	(D)	Formation of glass
Calor	ific value of kerosene is		
A)	1000 k.cal/kg	(B)	11,100 k.cal/kg
C)	12,200 k.cal/kg	(D)	9,200 k.cal/kg
LPG s	stands for		
(A)	Large Petrolium Gas	(B)	Low Petrolium Gas
(C)	Liquefied Petrolium Gas	(D)	Lime Petrolium Gas
Natur	al gas is generally associated with		
(A)	Soil deposits	(B)	Ore deposits
(C)	Stone deposits	(D)	Petrolium deposits
Oil ga	as is obtained by cracking of		
(A)	Coconut oil	(B)	Sunflower oil
(C)	Groundnut oil	(D)	Kerosene oil
Comb	oustion is a/anchemical	reacti	on.
(A)	Exothermic	(B)	Endothermic
(C)	Mesothermic	(D)	Deothermic
Γhe a	pparatus used for determining clo	ud poi	nt and pour point of crude oil is
(A)	Pour point apparatus	(B)	Bro point apparatus
(C)	Sour point apparatus	(D)	Glow point apparatus
	alor A) C) PG: AA) C) latur A) C) Comb A) C) the a A)	alorific value of kerosene is  A) 1000 k.cal/kg  C) 12,200 k.cal/kg  PG stands for  A) Large Petrolium Gas  C) Liquefied Petrolium Gas  (atural gas is generally associated with  A) Soil deposits  C) Stone deposits  bil gas is obtained by cracking of  A) Coconut oil  C) Groundnut oil  combustion is a/an chemical  A) Exothermic  C) Mesothermic  the apparatus used for determining clo  A) Pour point apparatus	alorific value of kerosene is  A) 1000 k.cal/kg  B)  C) 12,200 k.cal/kg  C) 12,200 k.cal/kg  C) 12,200 k.cal/kg  C)  PG stands for  A) Large Petrolium Gas  C) Liquefied Petrolium Gas  D)  Intural gas is generally associated with  A) Soil deposits  C) Stone deposits  C) Stone deposits  D)  Intural gas is obtained by cracking of  A) Coconut oil  C) Groundnut oil  C) Groundnut oil  C) Groundnut oil  C) Mesothermic  C) Mesothermic  C) Mesothermic  C) Mesothermic  C)  The apparatus used for determining cloud point  A) Pour point apparatus  (B)

142.	Cnro	mite retractories are				
	(A)	Acidic refractories			(B)	Basic refractories
	(C)	Neutral refractories			(D)	None of these
143.	Fish	sealing and chips like	human	nails a	are she	own on enamel surface because of
	(A)	Under firing			(B)	Faulty composition
	(C)	Over firing			(D)	Old raw materials
144.	Yelle	ow colour enamel is be	est mad	e by a	dding	1 to 3% of
	(A)	Copper oxide			(B)	Iron oxide
	(C)	Zinc oxide			(D)	Uranium oxide
145.	Hard	lening of lacquer is ma	ainly du	ue to o	xidati	on of lacquer is carried by an enzyme called
	(A)	Lactose			(B)	Glucose
	(C)	Glygyrine			(D)	Pituitaries
146.	Dipp	oing method of enamel	coatin	g appl	icatio	n is sometime known as
	(A)	Pouring			(B)	Sparing
	(C)	Slushing			(D)	Sprinkling
147.	Chr	omium oxide gives		colou	r in hi	gh B <sub>2</sub> O <sub>3</sub> glaze.
	(A)	Black			(B)	Blue
	(C)	Green			(D)	Yellow
148.	The	thickness of glaze coa	ting on	ceram	nic wa	re should be
1104	(A)	Thin				Very thin
	(C)	Thick			` ′	Very thick
_					-	1 887 - 1

149.	Addi	tion of	makes	the fruit mill	cy w	hite.	
	(A)	NaCl		(	B)	KCl	
	(C)	CaO		(	D)	MgO	
150.	The	glaze defec	ct peeling is car	used due to			
	(A)	High ther	mal expansion	of glaze			
	(B)	Low then	mal expansion	of glaze			
	(C)	High plas	sticity in the gl	aze			
	(D)	Low plas	ticity in the gla	aze			
151.	The	purpose of	calcining Zinc	oxide is to			
	(A)	vaporize	Iron and other	impurities			
	(B)	eliminate	Carbon di-oxi	de			
	(C)	make it b	rittle				
	(D)	make it is	nsoluble				
152.	The	essential d	ifference betw	een a glaze a	ınd a	a engobe is	
	(A)	More glo	ssy phase in gl	aze			
	(B)	Less glos	ssy phase in gla	aze			
	(C)	Equal glo	ossy phase in b	oth			
	(D)	More glo	ossy phase in er	ngobe			
153.	Sani	tary pipes	are shaped by				
	(A)	Extrusion	n		(B)	Jiggering	
	(C)	Jollying		(	(D)	Screw Press	
154.	Whi	ch one of t	the following c	ontrols the t	hick	ness of plate in tape casting	ng?
	(A)	Blade			(B)	Viscosity of the slurry	
	(C)	Drying s	hrinkage		(D)	All of the above	
				2 E	D	rob Work	

		Space Fo	or Rot	igh Work		
	(C)	20 mm	(D)	25 mm		
	(A)	10 mm	(B)	15 mm		
160.	In so	oundness test cement should not exp	and n	nore than		
	(C)	Rotary	(D)	Chamber		
	(A)	Tunnel	(B)	Mangle		
159.		Drier is used for drying of mat	егials			
	(~)		` /			
	(C)	Electrical	(D)	Mechanical		
2001	(A)	Centrifugal	(B)	Centripetal		
158	Cvcl	one separator works on pr	incipl	e.		
	(C)	Water consistency	(D)	Soundness		
	(A)	Initial setting time	(B)	Final setting time		
157.	Annı	ular collar needle is used to test				
	(C)	Three directions	(D)	All directions		
	(A)	One direction	(B)	Two directions		
156.	In Iso	o-Static press the pressure is applied	l from			
	(C)	Sciew conveyor	(D)	The distriction of the second		
		Screw conveyor	(D)	Pneumatic conveyor		
133.	(A)	Apson conveyor	(B)	Belt conveyor		
l55.	Which is the most suitable conveyor for transportation of sticky material:					

161.	In the	e manufacturing of Bari	um titanate	capac	itors the % of BaCO <sub>3</sub> is
	(A)	50%		(B)	44%
	(C)	52%		(D)	40%
162.	Whe as	n insulator transforms p	oart of elect	rical e	nergy into heat, the loss of energy is known
	(A)	Dielectric strength		(B)	Power factor
	(C)	Dielectric wastage		(D)	Heat factor
163.	PLZ	T is used as			
	(A)	Capacitor		(B)	Resistor
	(C)	Piezoelectricity		(D)	Ferrite
164.		is used in internal	combustio	n engi	nes.
	(A)	Honey comb		(B)	Filters
	(C)	Semi conductor		(D)	Spark plug
165.	The	chemical formula of ma	ignet is		
	(A)	Fe <sub>3</sub> O <sub>4</sub>		(B)	Fe
	(C)	FeO		(D)	Fe <sub>2</sub> O
166.	Whi	ich one of the following	offers less	resista	nce to the flow of current?
	(A)	Insulators		(B)	Conductors
	(C)	Semi conductors		(D)	Ionic conductors
167.	The	fracture toughness of T	ungsten car	bide c	utting tool is
	(A)	6.6 MPa		(B)	10.5 MPa
	(C)	13.8 MPa		(D)	15 MPa

168.	Tiles	are usually shaped by			
	(A)	Plastic pressing	(B)	Dry pressing	
	(C)	Casting	(D)	Extrusion	
160	Defl	occulant increases of the s	lin		
107.		Fluidity	(B)	Density	
	(A)	Viscosity	(D)	Thixotropy	
	(C)	Viscosity	(D)	Imadiopy	
170.	Tem	perature Variations are detected by		sensor.	
	(A)	Gas	(B)	Humidity	
	(C)	Oxygen	(D)	Thermo	
171.	Clay	is a			
	(A)	Primary rock	(B)	Secondary rock	
	(C)	Tertiary rock	(D)	Sigma rock	
172.	Pres	ence of alkalies makes the body			
	(A)	Hard	(B)	Compact	
	(C)	Porous	(D)	Crystalline	
			111	A Assure	
		mina forms dimensional crysta			
	(A)	one		two	
	(C)	three	(D)	four	
174	Tha	atable form of cilian at room tempe	rature	10	
1/4.		stable form of silica at room tempe	(B)	Beta-Quartz	
	(A)	Alpha-Quartz	(D)	Tetra-Quartz	
	(C)	Gama-Quartz	(1)	10110-Anne	

175.	Exan	aple for organic binder is						
	(A)	Gelatin	(B)	Dextrine				
	(C)	Gum arabic	(D)	All these				
176.	Carb	ides are						
1.00	(A)	Very hard material	(B)	Very soft material				
	(C)	Very brittle material	(D)	Plastic material				
177.	Nitri	des are the bodies which shows						
	(A)	High refractoriness	(B)	Low refractoriness				
	(C)	Medium refractoriness	(D)	High thermal spalling				
178.	Bori	des belongs to						
	(A)	Transition metals	(B)	Non-metals				
	(C)	Noble metals	(D)	Carbon				
179.	Silic	side is a compound which has						
1770	(A)	Silicon with more electropositive	eleme	ent				
	(B)	Silicon with more electronegative						
	, ,							
	(D)	<ul><li>(C) Silicon with less electropositive element</li><li>(D) Silicon with less electronegative element</li></ul>						
180.	Che	mical formula of Titanium di borid	e is					
	(A)	TiB	(B)	TiB <sub>2</sub>				
	(C)	Ti <sub>2</sub> B	(D)	Ti <sub>3</sub> B				
		Space F	or Ro	ugh Work				

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